

REPORT DOCUMENTATION PAGEForm Approved
OMB No. 0704-0188

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1. REPORT DATE (DD-MM-YYYY)

28-02-03

2. REPORT TYPE

View Graphs

3. DATES COVERED (From - To)**4. TITLE AND SUBTITLE**

Organic Polymers Modified with Inorganic Polyhedra

5a. CONTRACT NUMBER

F04611-99-C-0025

5b. GRANT NUMBER**5c. PROGRAM ELEMENT NUMBER****6. AUTHOR(S)**

Timothy S. Haddad, Rene Gonzalez

5d. PROJECT NUMBER

2303

5e. TASK NUMBER

M1A3

5f. WORK UNIT NUMBER**7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES)**ERC, Inc.
10 E. Saturn Blvd.
Edwards AFB, CA 93524-7680**8. PERFORMING ORGANIZATION
REPORT NUMBER****9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES)**Air Force Research Laboratory (AFMC)
AFRL/PRS
5 Pollux Drive
Edwards AFB CA 93524-7048**10. SPONSOR/MONITOR'S
ACRONYM(S)****11. SPONSOR/MONITOR'S
NUMBER(S)**

AFRL-PR-ED-VG-2003-048

12. DISTRIBUTION / AVAILABILITY STATEMENT

Approved for public release; distribution unlimited.

13. SUPPLEMENTARY NOTES**14. ABSTRACT**

20031003 092

15. SUBJECT TERMS**16. SECURITY CLASSIFICATION OF:****a. REPORT**

Unclassified

b. ABSTRACT

Unclassified

c. THIS PAGE

Unclassified

**17. LIMITATION
OF ABSTRACT**

A

**18. NUMBER
OF PAGES****19a. NAME OF RESPONSIBLE
PERSON**

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19b. TELEPHONE NUMBER(include area code)
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FILE

MEMORANDUM FOR PRS (In-House/Contractor Publication)

FROM: PROI (STINFO)

28 Feb 2003

SUBJECT: Authorization for Release of Technical Information, Control Number: **AFRL-PR-ED-VG-2003-048**
Timothy S. Haddad and Capt. Rene Gonzalez, "Organic Polymers Modified with Inorganic Polyhedra"

American Chemical Society Conference
(New Orleans, LA, 23-27 Mar 2003) Deadline: 21 Mar 2003

(Statement A)

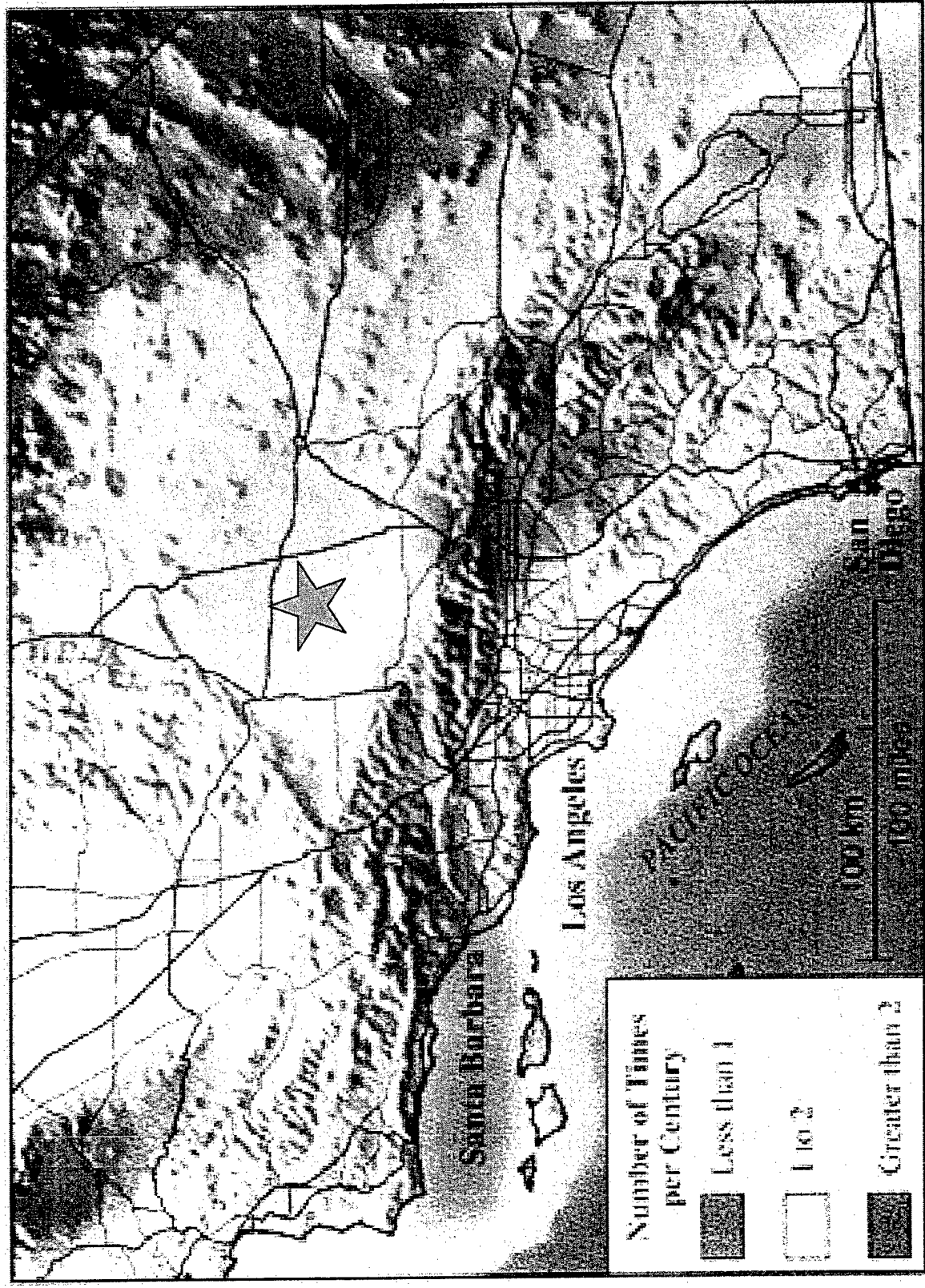


ORGANIC POLYMERS MODIFIED WITH INORGANIC POLYHEDRA

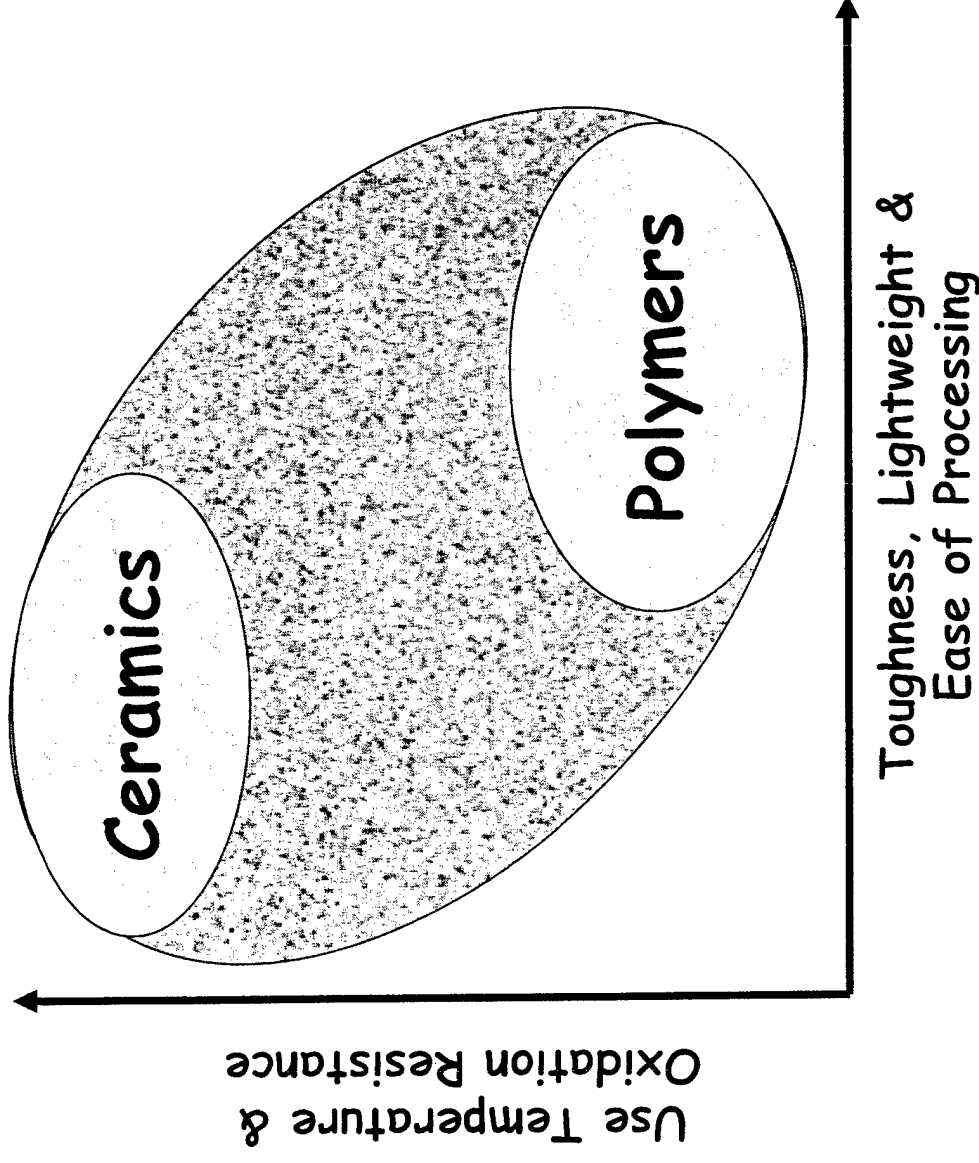
**Tim Haddad and Rene Gonzalez
ERC Inc., Air Force Research Lab**

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Edwards Air Force Base, CA



Hybrid Inorganic/Organic Polymers

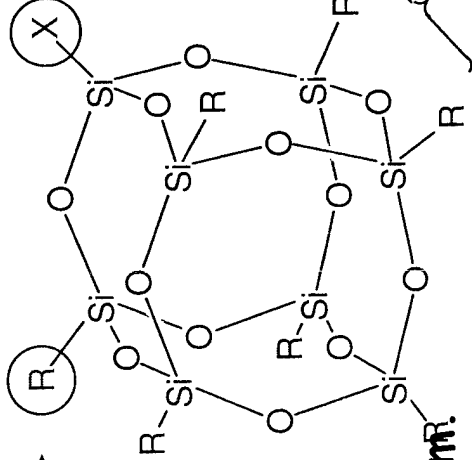


• Hybrid plastics bridge the differences between ceramics and polymers

Anatomy of a Polyhedral Oligomeric Silsesquioxane (POSS) Macromer

Nonreactive organic (R) \longrightarrow groups for solubilization and compatibilization.

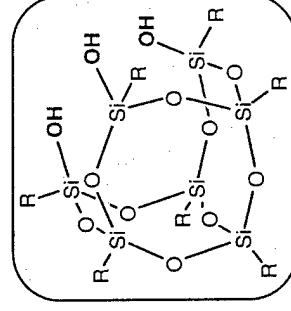
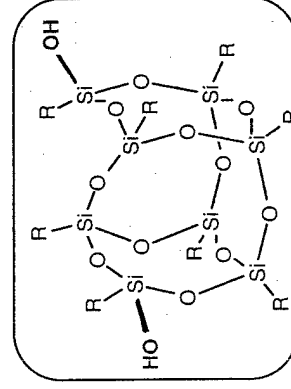
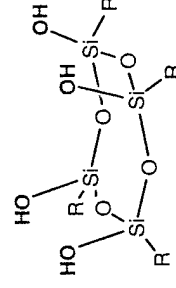
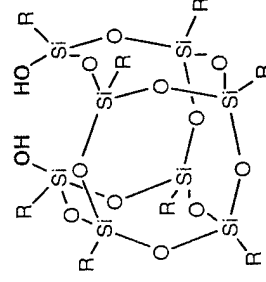
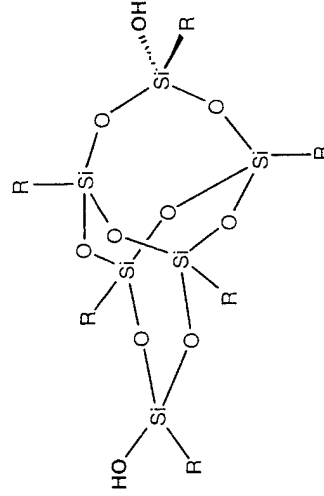
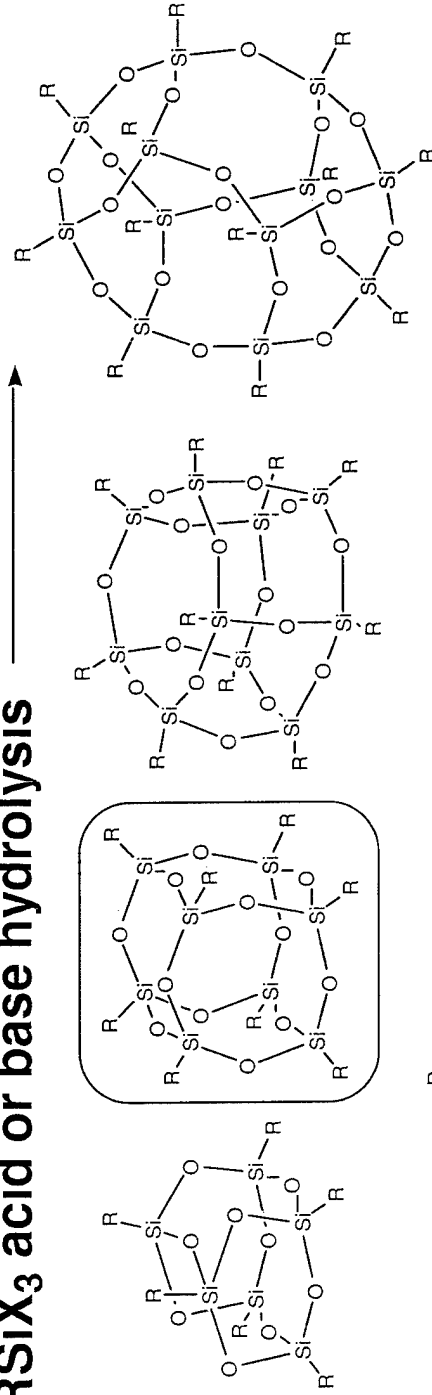
Nanoscopic in size with an Si-Si distance of 0.5 nm and a R-R distance of 1.5 nm.



Precise three-dimensional structure for molecular level reinforcement of polymer segments and coils.

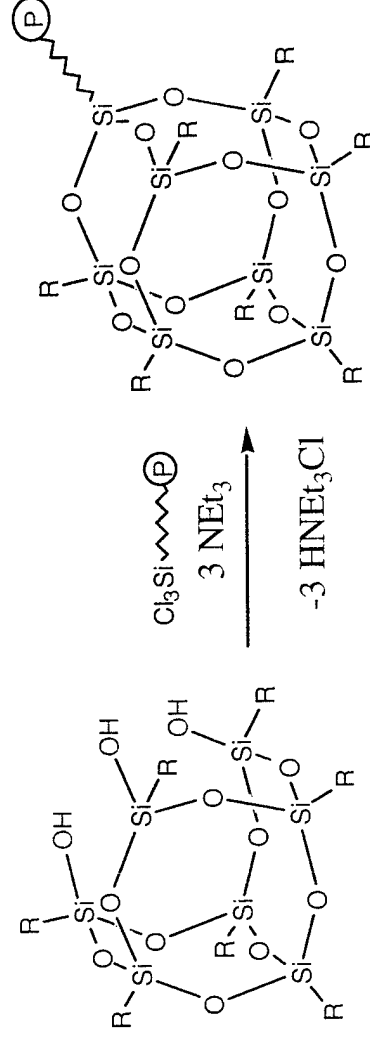
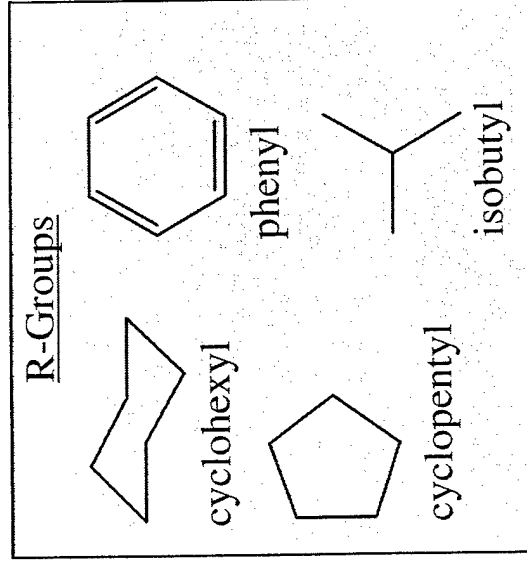
POSS Synthesis

RSiX₃ acid or base hydrolysis



Brown & Vogt: JACS, 1965, 4313
 Feher et al: JACS, 1989, 1741;
 Organometallics, 1991, 2526;
 Chem Comm, 1999, 1705, 2309

POSS Macromers For Nanocomposites

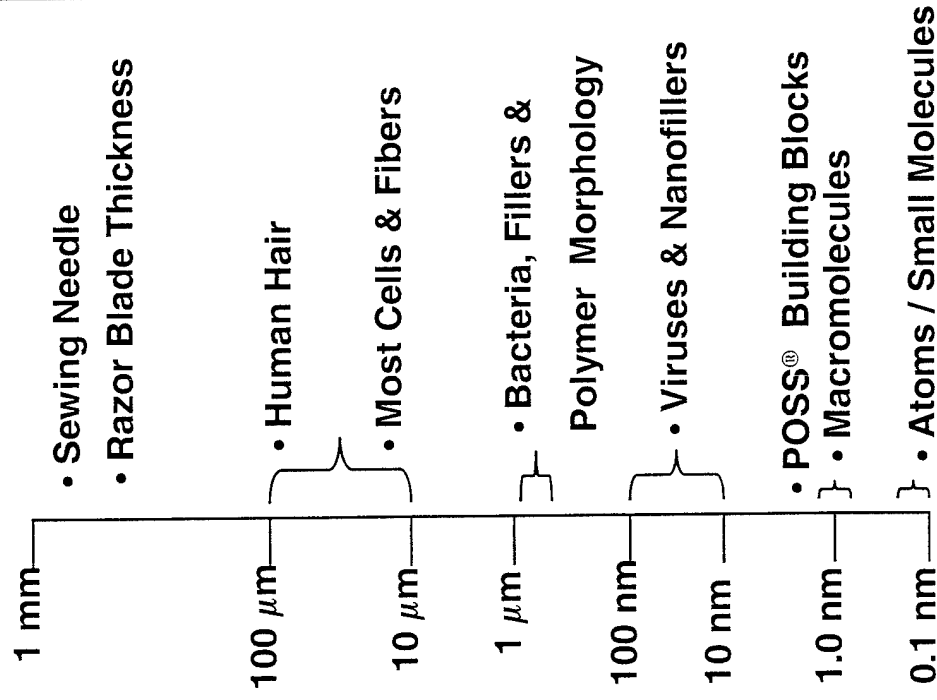


Halides	Nitriles	Silanes	Styryls
Alcohols	Amines	Silanols	α -olefins
Esters	Isocyanates	Silylchlorides	Acrylics
Bisphenols	Epoxides		Norbornenyls

POSS-based macromers are available through either **Geleste** or **Aldrich**

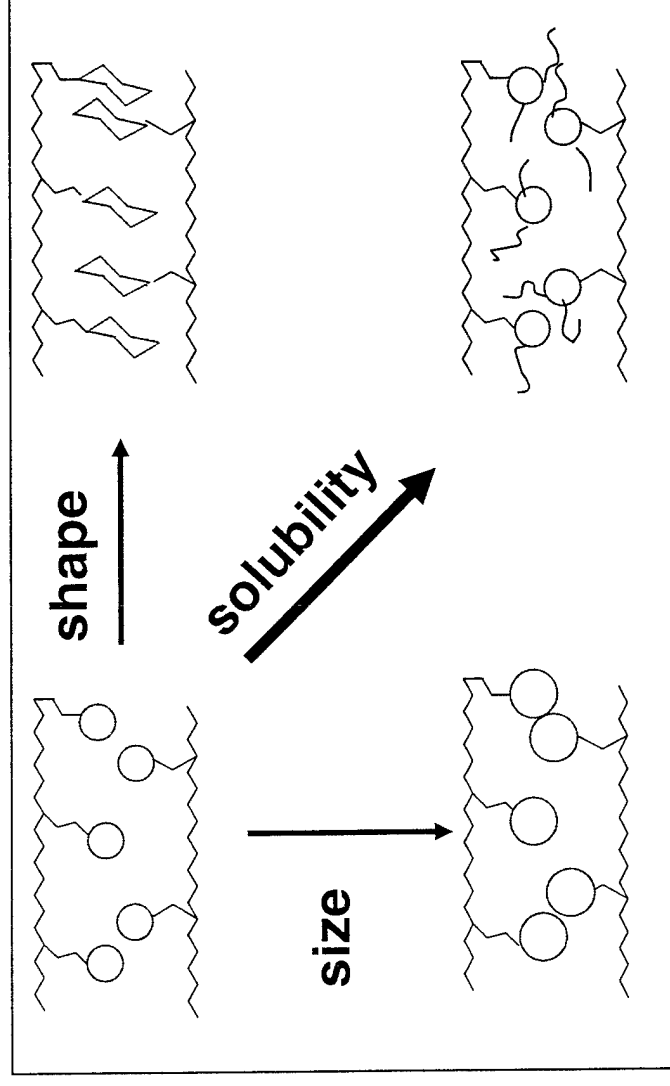
POSS technology is commercialized by **Hybrid Plastics** in Fountain Valley CA

Why POSS and Why Nano?



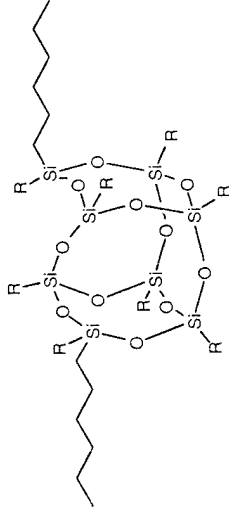
Field	Property	Critical Length
Electronics	Tunneling	1-100 nm
Optical	Quantum Well	1-100 nm
	Wave Decay	10-1000 nm
Polymers	Primary Structure	0.1-10 nm
	Secondary Structure	10-1000 nm
Mechanics	Dislocation Interaction	1-1000 nm
	Crack Tip Radius	1-100 nm
	Entanglement Rad.	10-50 nm
Therm-Mech.	Chain Motion	0.5-50 nm
Nucleation	Defect	0.1-10 nm
	Critical Nucleus Size	1-10 nm
	Surface Corrugation	1-10 nm
Catalysis	Surface Topology	1-10 nm
Biology	Cell Walls	1-100 nm
Membranes	Porosity Control	0.1-5 nm

Structure-Property Relationships

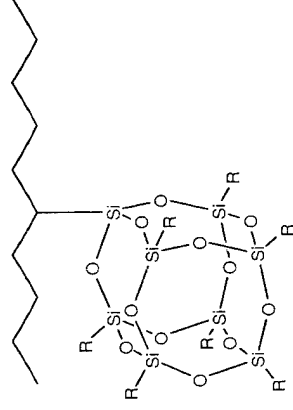


- Maximizing property enhancements through changes at the nano level
- Polymer miscibility vs. POSS/POSS interactions

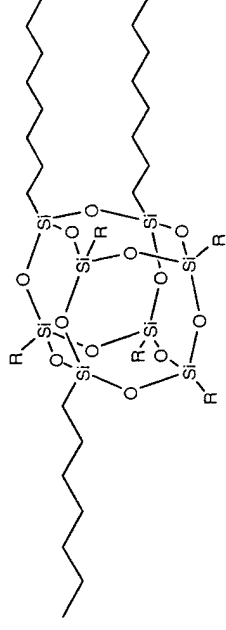
POSS Polymer Incorporation



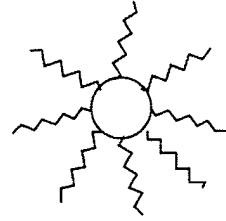
POSS Bead



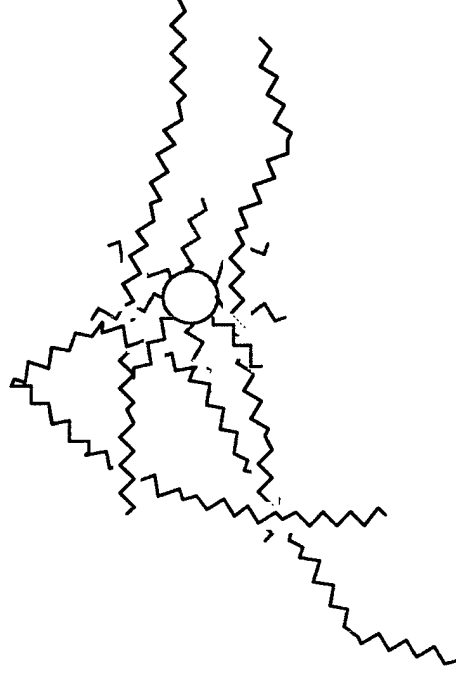
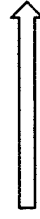
POSS Pendant



POSS Crosslinking

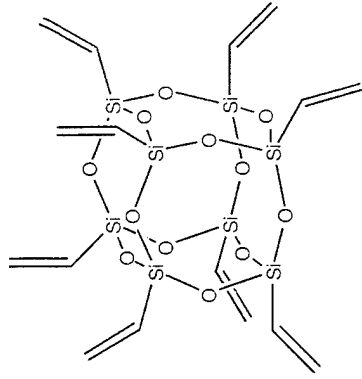


POSS Blending

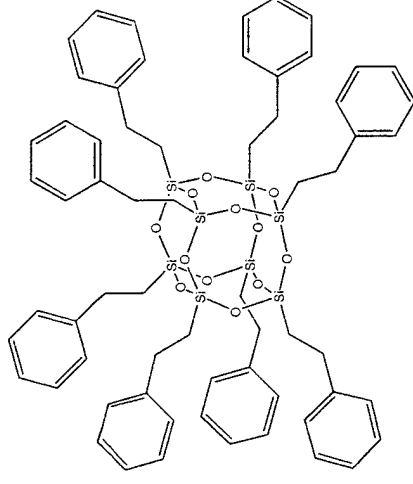


Size & Shape

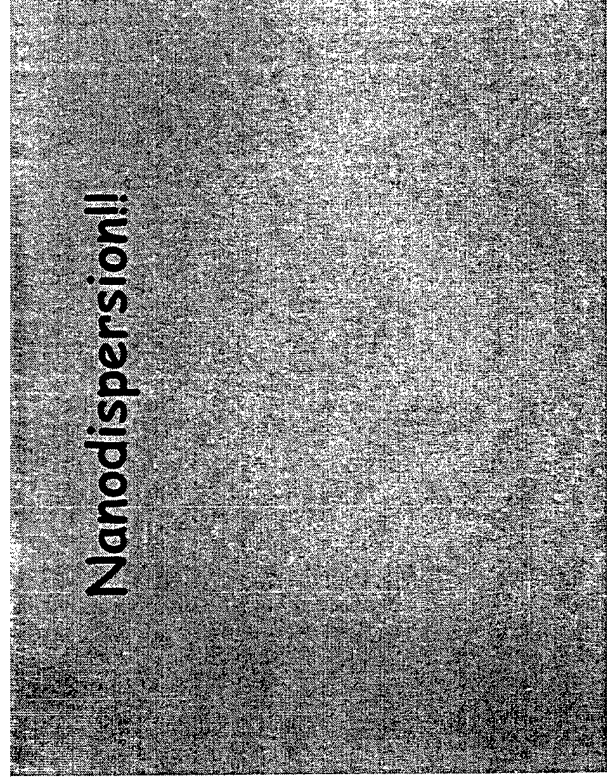
50 Wt % POSS Blends in 2 Million MW PS



Vi₈T₈



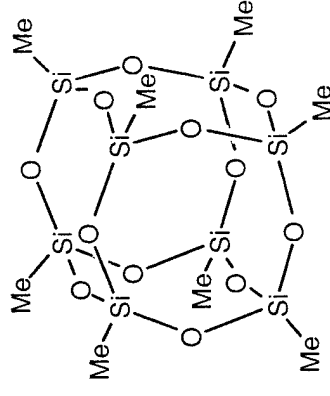
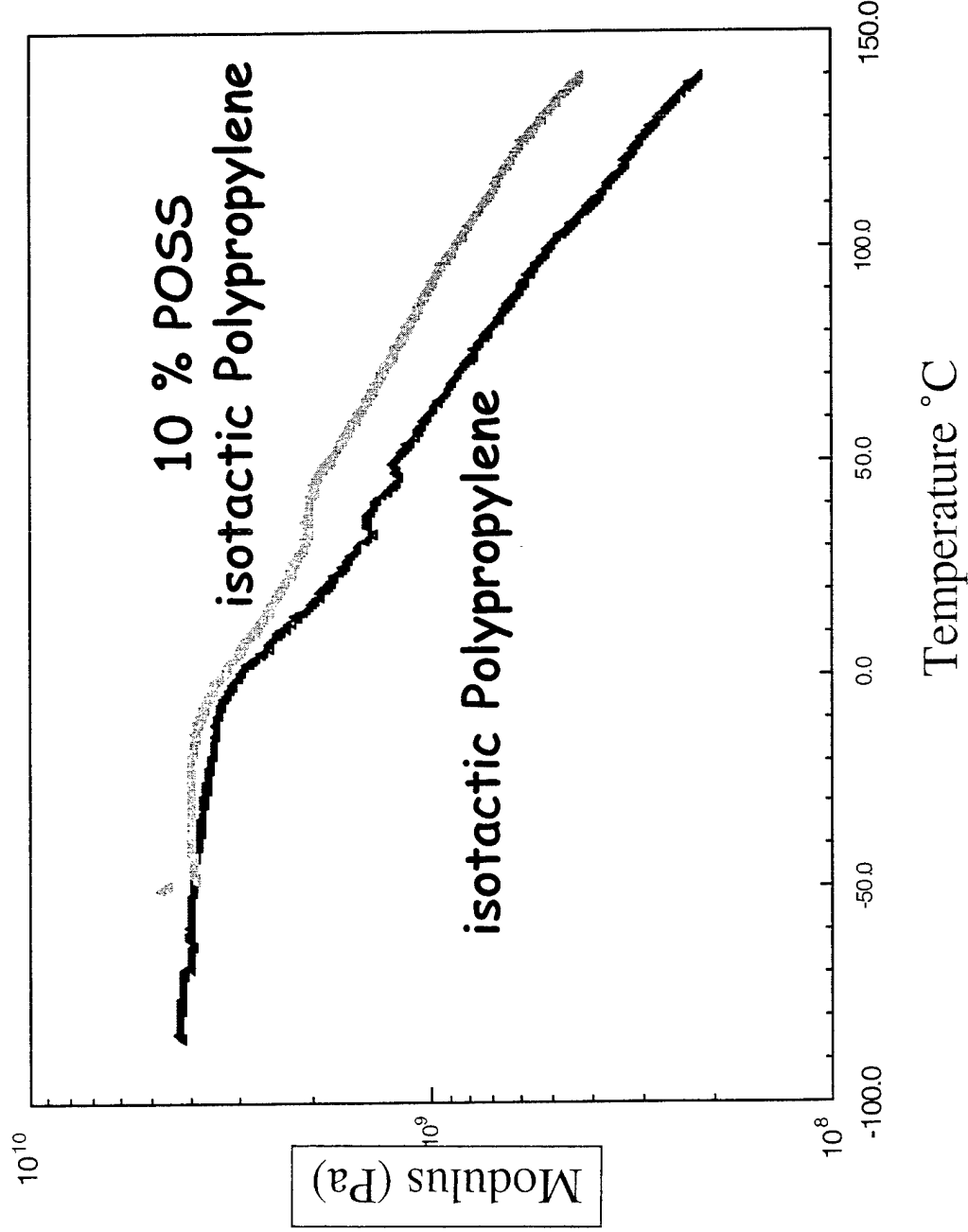
Phenethyl₈T₈



Nanodispersion!!

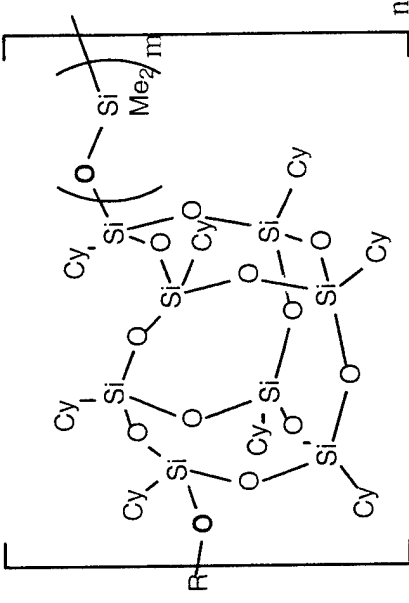
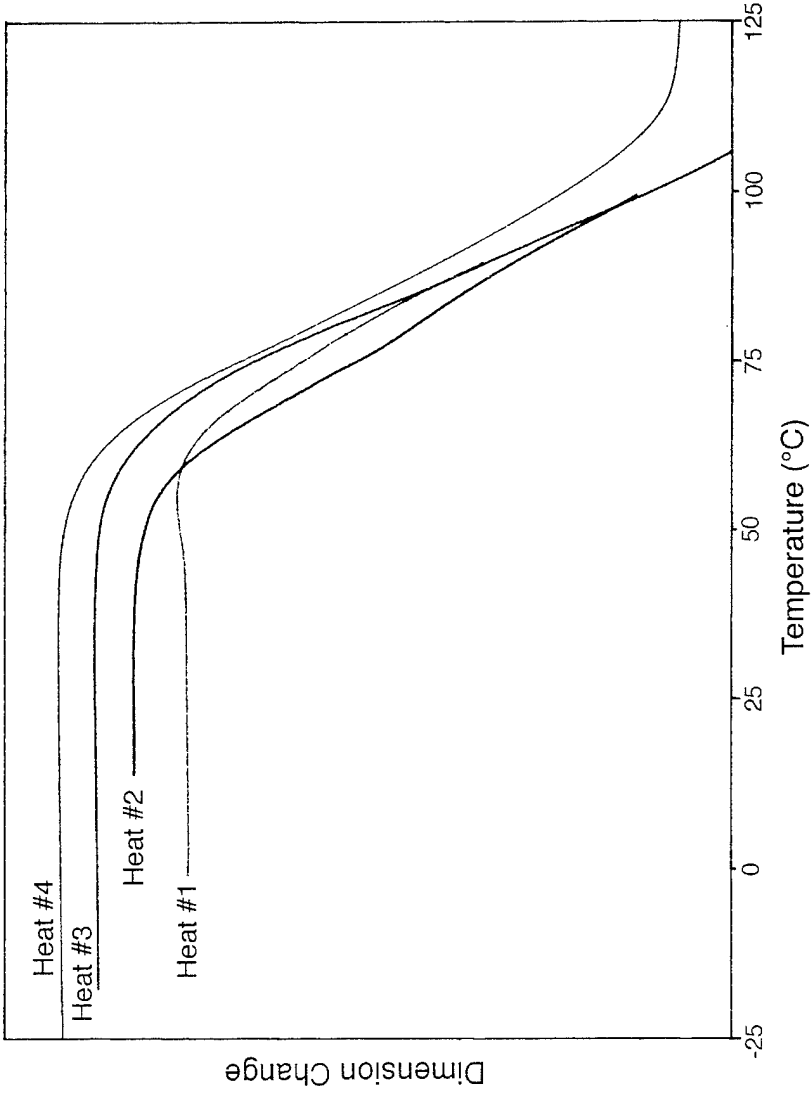
1 μm

DMA of 10 Wt % POSS in isotactic Polypropylene

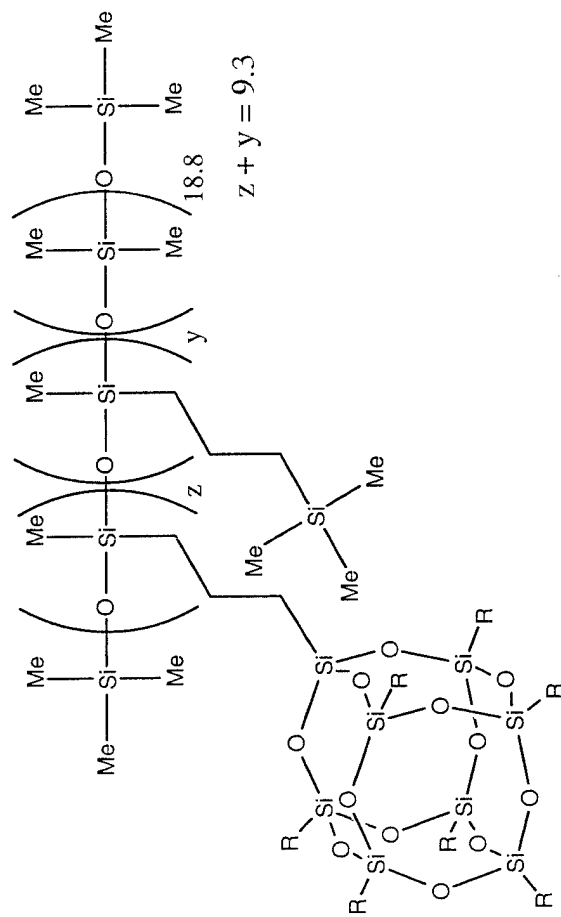
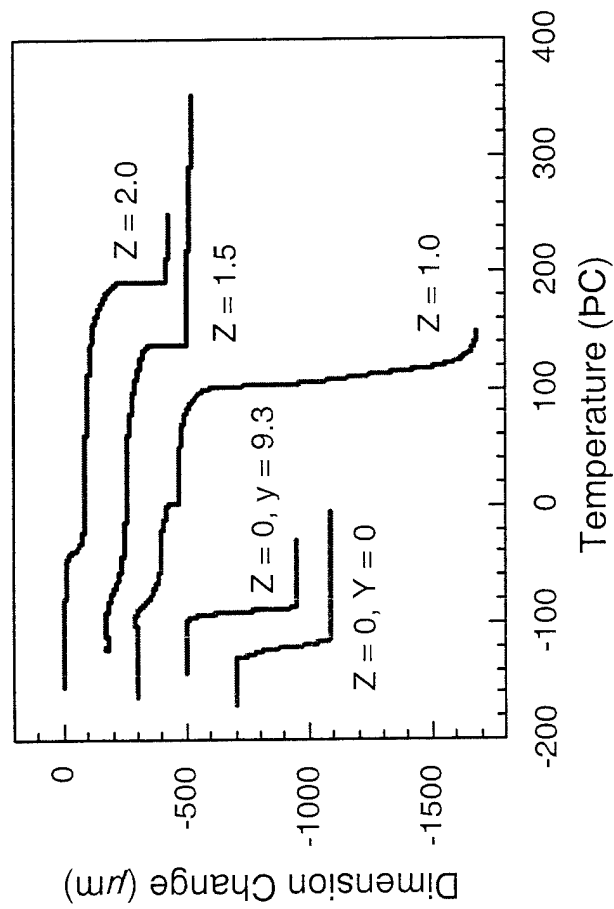


PDMS-POSS TMA Characterization

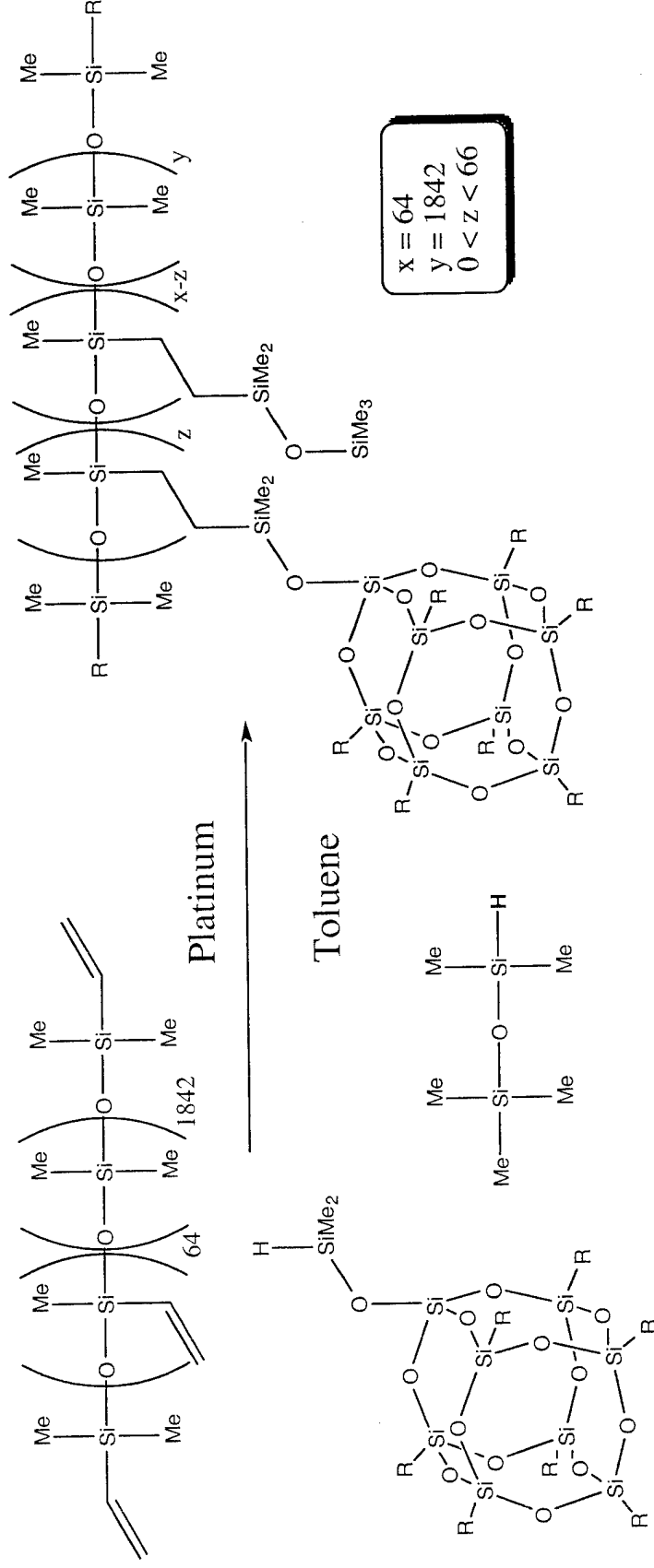
The POSS/Siloxane copolymers with four or more Si-O repeat units in the siloxane segment have softening temperatures well below the decomposition temperatures.



TMA of Pendent POSS-Siloxanes



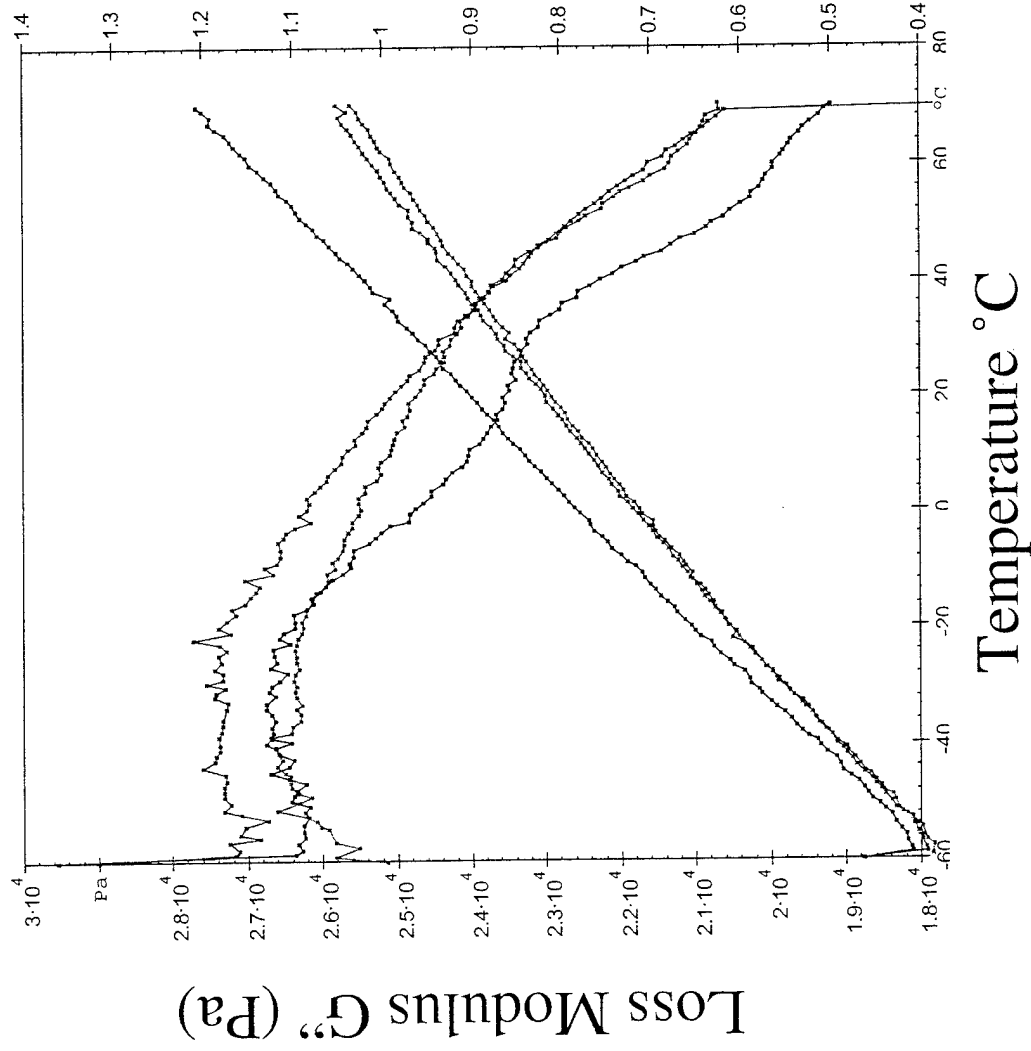
Hydrosilation to High MW PDMS



There are about 7 POSS-
macromers per PDMS chain

Used 5 weight % POSS

Comparison of Three T8-POSS Macromers

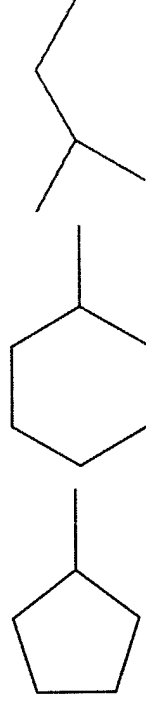
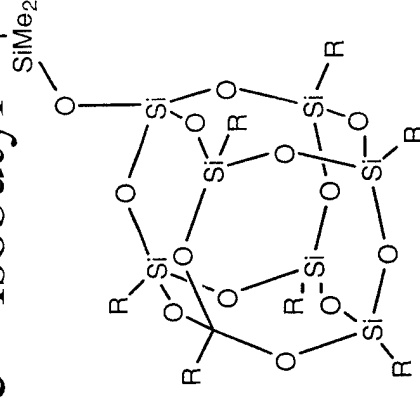


PDMS + 5 wt % POSS

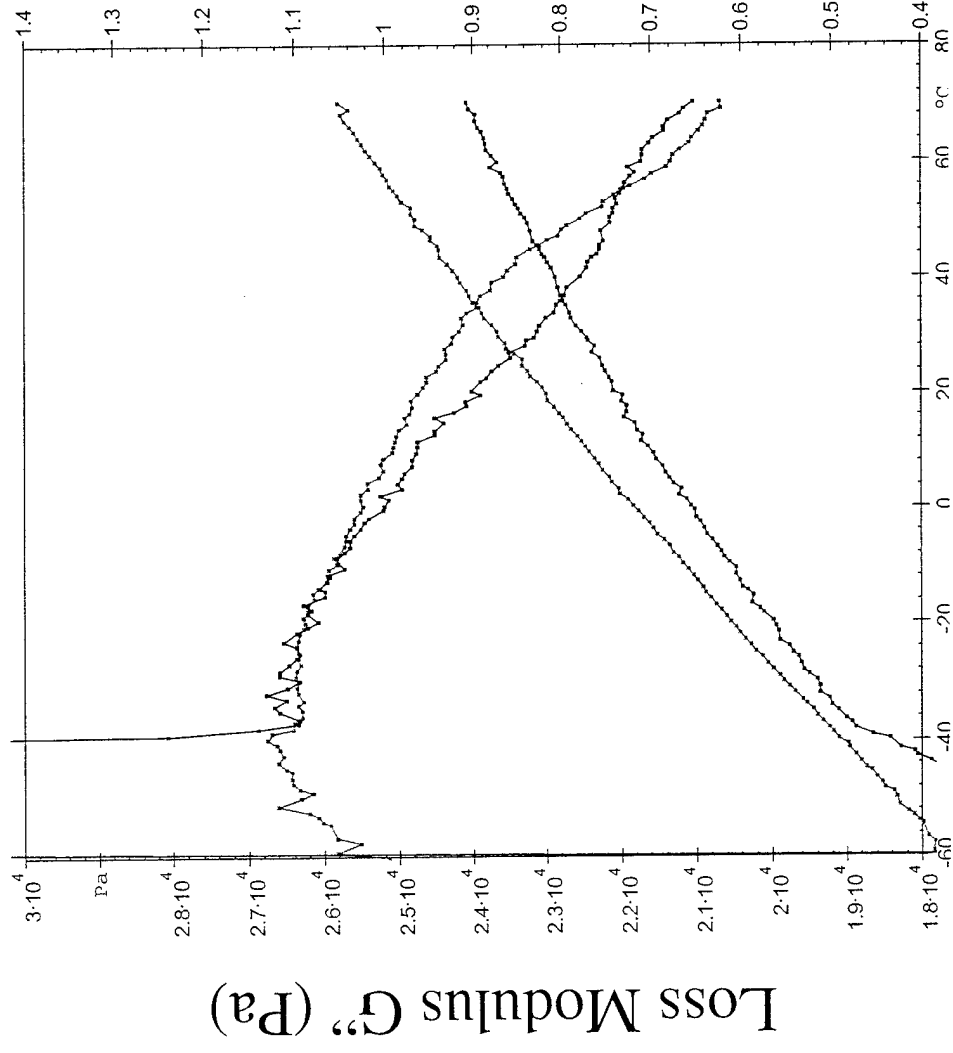
Blue = cyclopentyl

Red = cyclohexyl

Purple = isobutyl

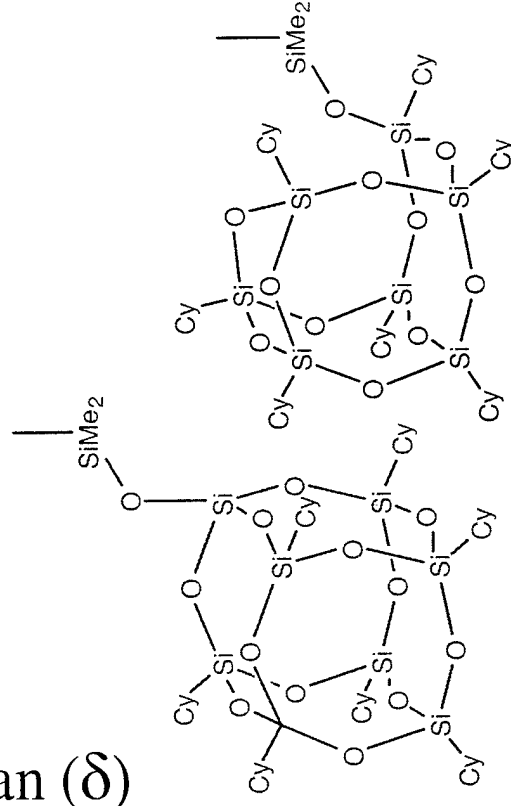


Comparison of Two POSS Polyhedra



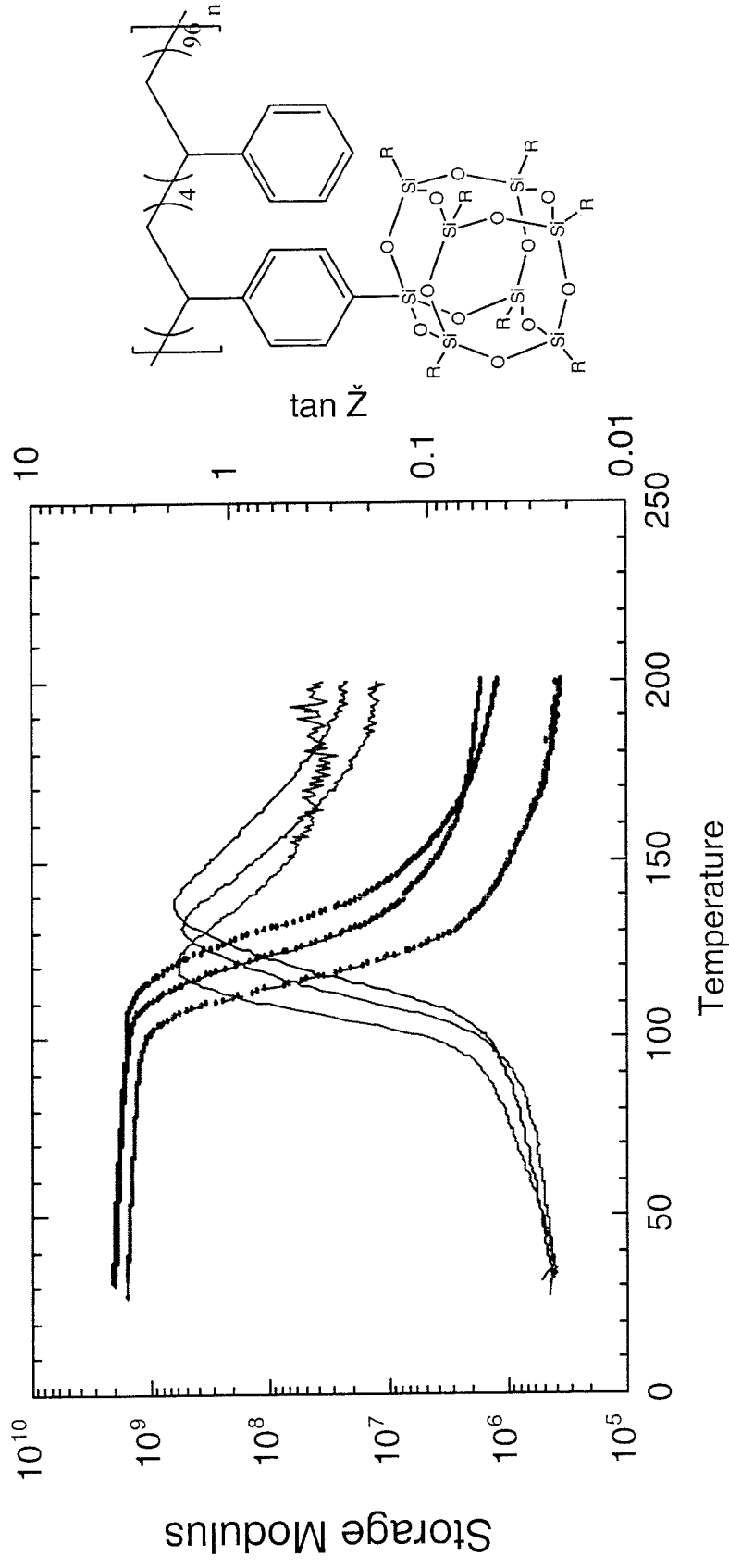
PDMS + 5 wt %
CyclohexylPOSS
Red = T8-POSS
Blue = T7-POSS

$\tan(\delta)$



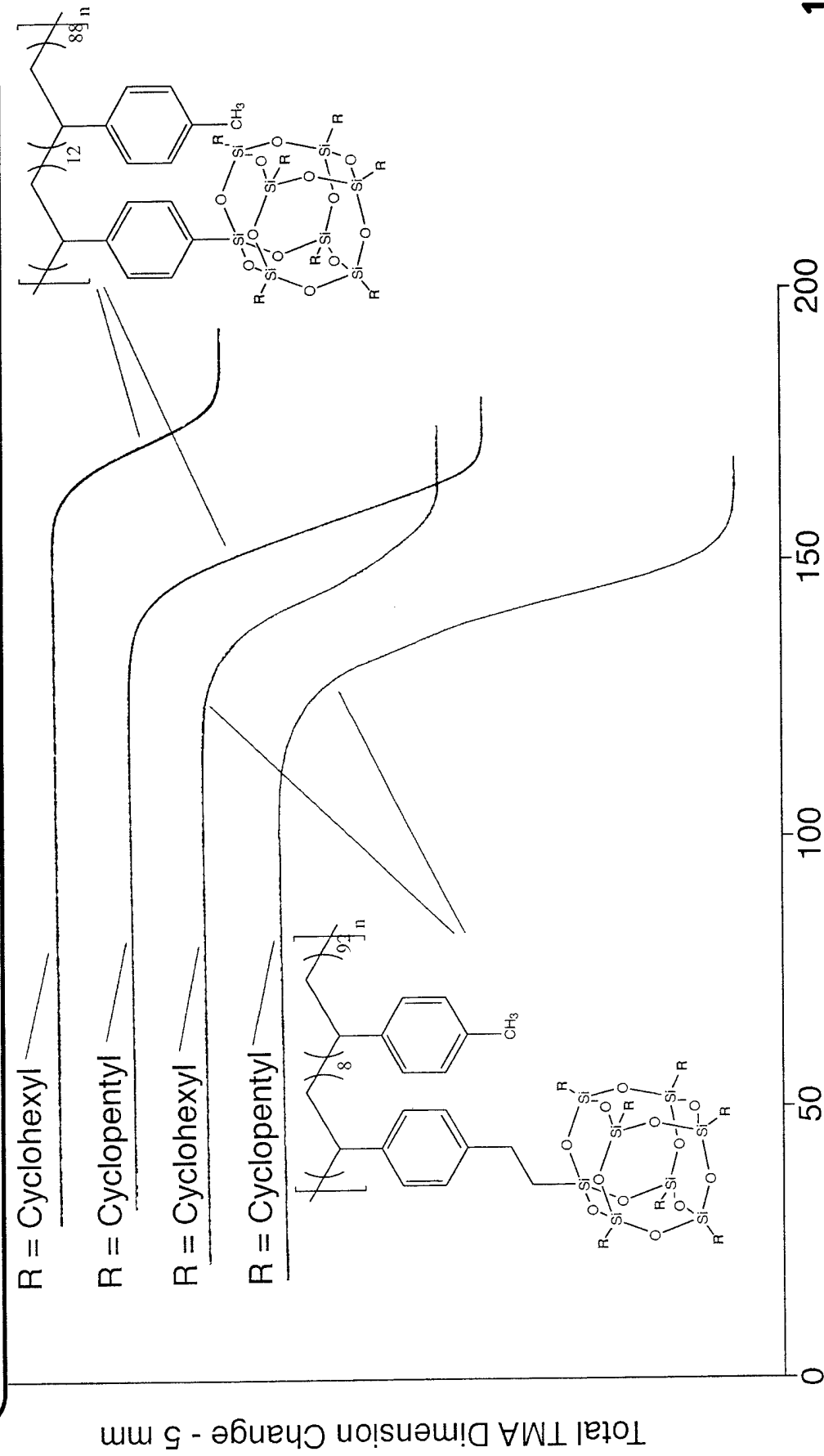
Temperature $^{\circ}\text{C}$

DMA of 30 wt % POSS Polystyrenes

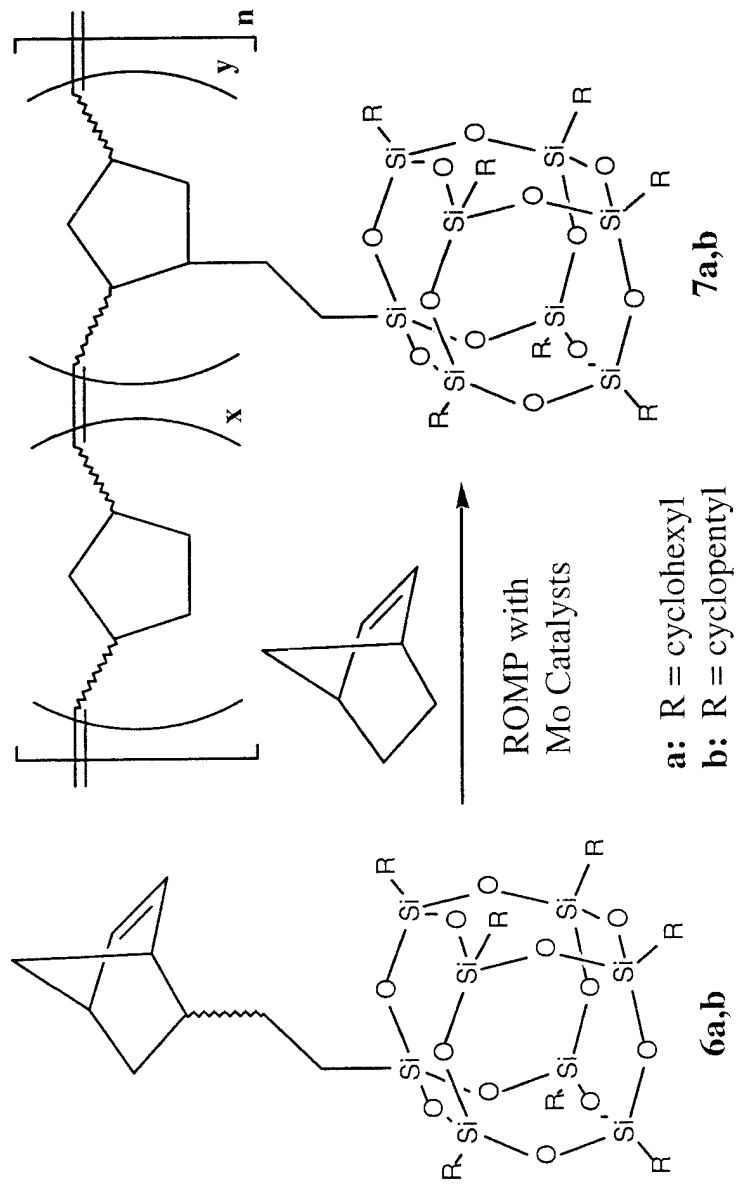


- Comparison of isobutyl, cyclopentyl & cyclohexyl
- Bulk polymerized samples

TMA Plot Comparison For POSS-Styryl and POSS-EthylStyryl Polymers (R = Cyclohexyl and Cyclopentyl)



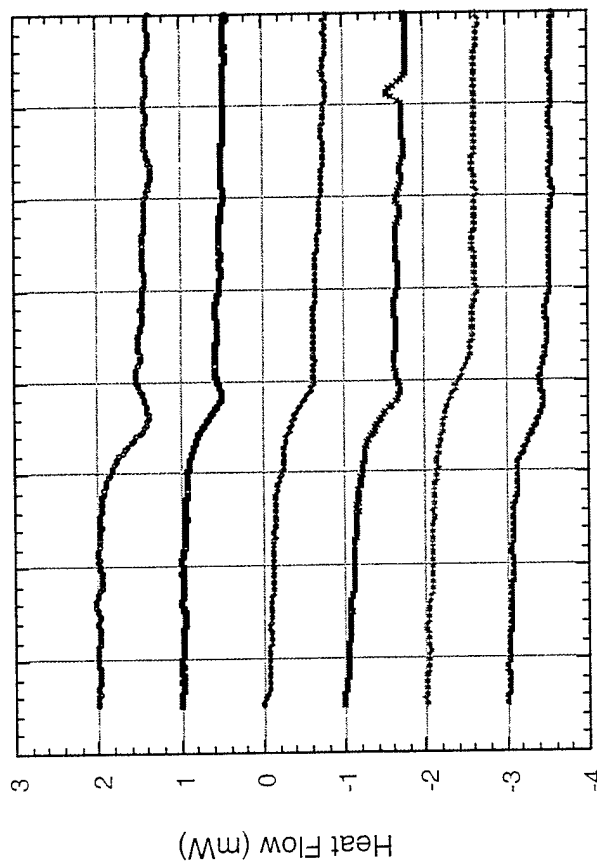
Polymerization of POSS Norbornenes



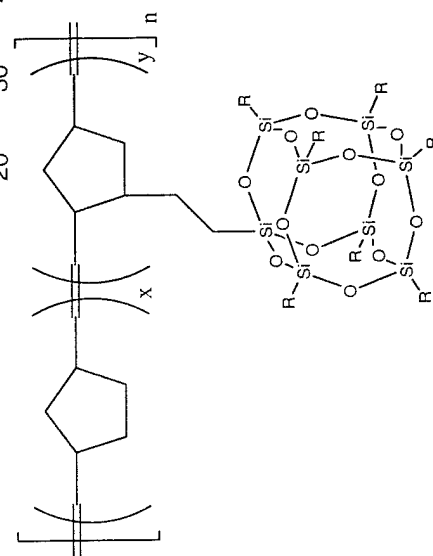
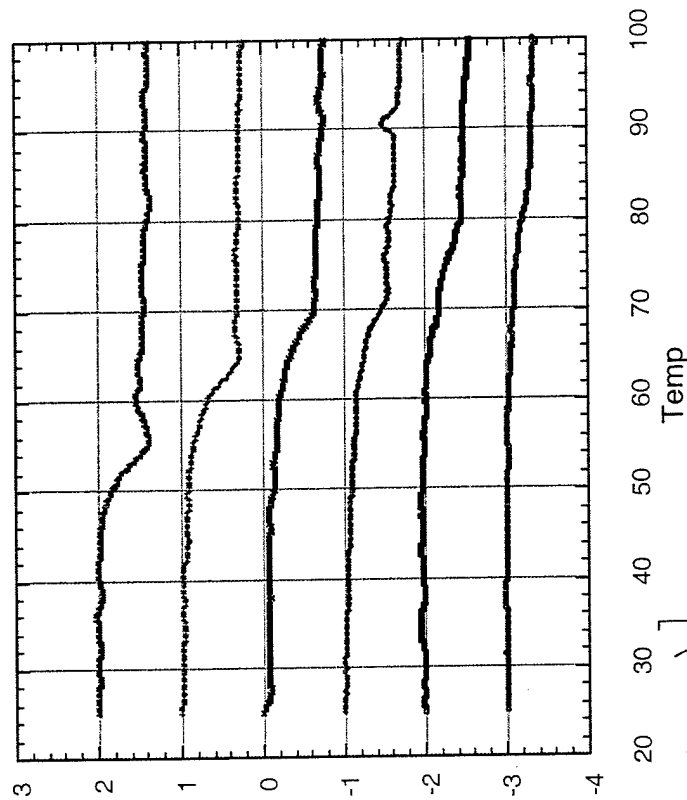
Both block and random copolymers were synthesized.
The wt. % POSS was varied from 0 to 50 wt. % POSS.

DSC Data for POSS-Norbornenes

CyNorb(0-50)-block

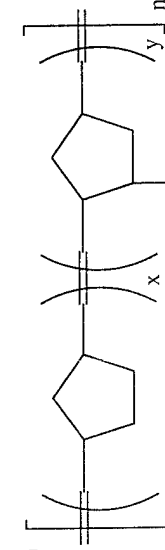
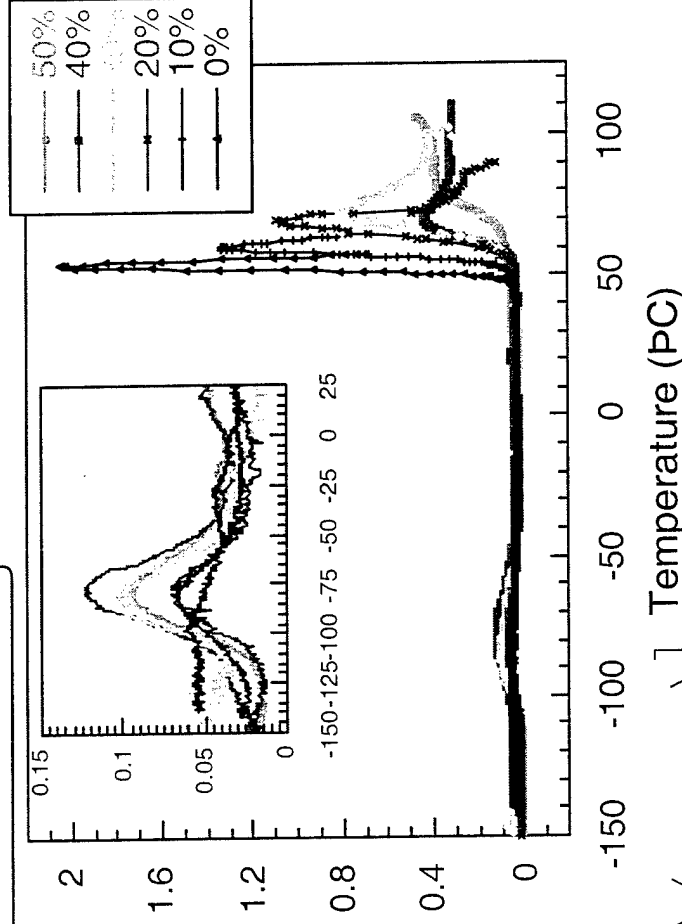
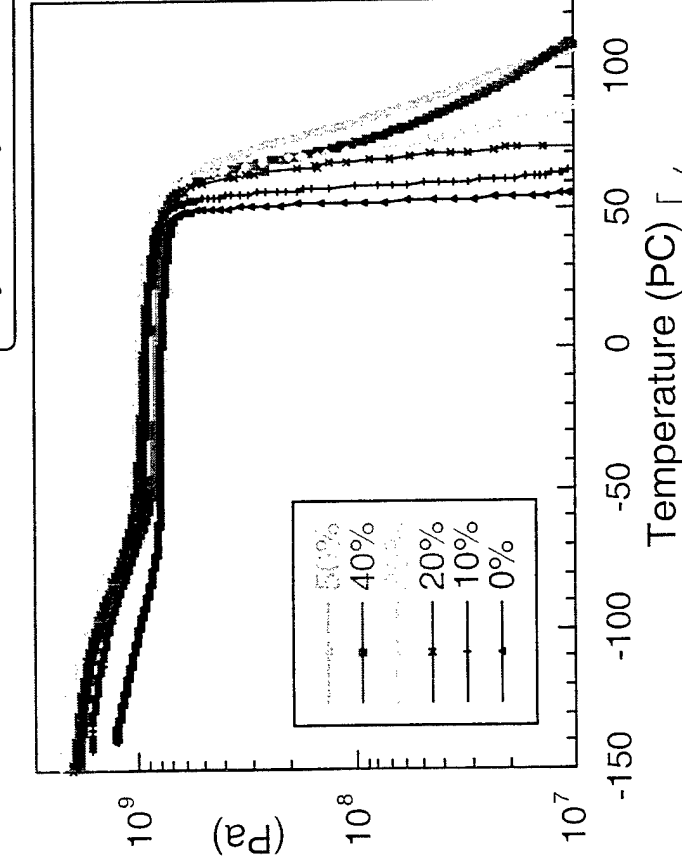


CyNorb(0-50)-random



Storage Modulus and Loss Tangent

Cyclohexyl Relaxation: 14.7 kcal/mol

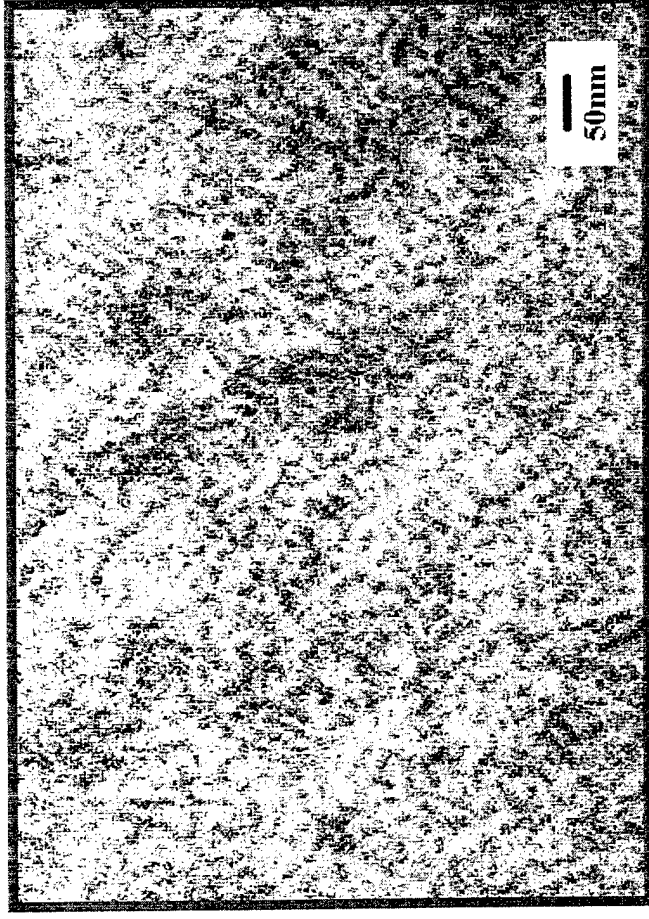


No Maximum for 50% CyPOSS

Various Wt % Cyclohexyl POSS Polynorbornene Random Copolymers

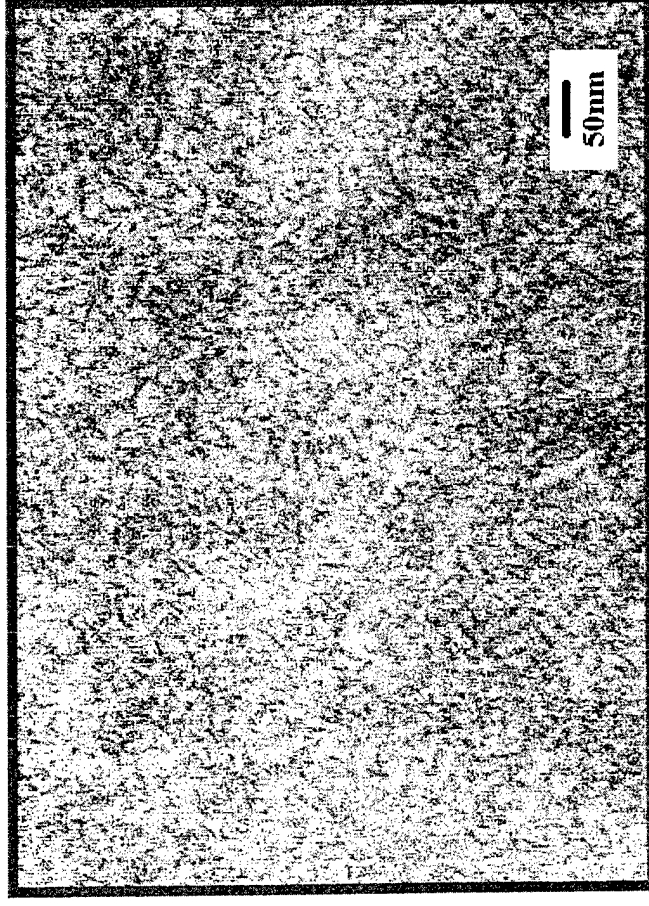
TEM of Random POSS Norbornenes

50CyPOSS/PN



"Coarse" Cylinder Nanostructure
(Diameter ~ 12nm)

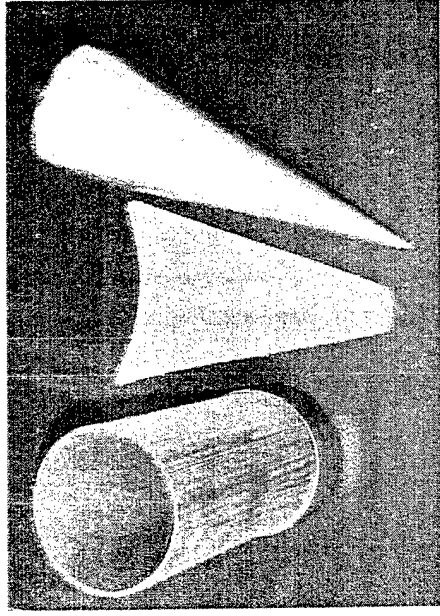
50CpPOSS/PN



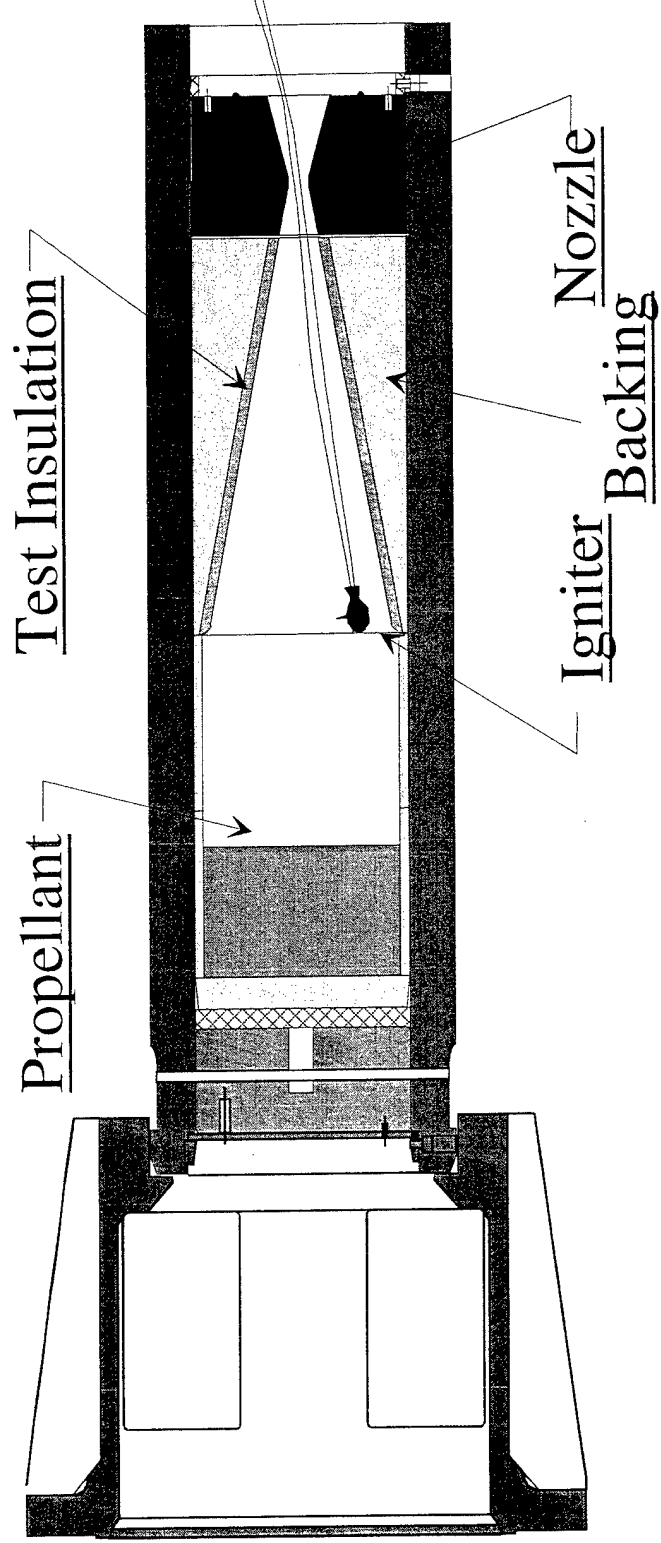
"Fine" Cylinder Nanostructure
(Diameter ~ 6nm)

CyclohexylPOSS-rich domains may entrain more unoriented polynorbornene chains than CyclopentylPOSS-rich domains.

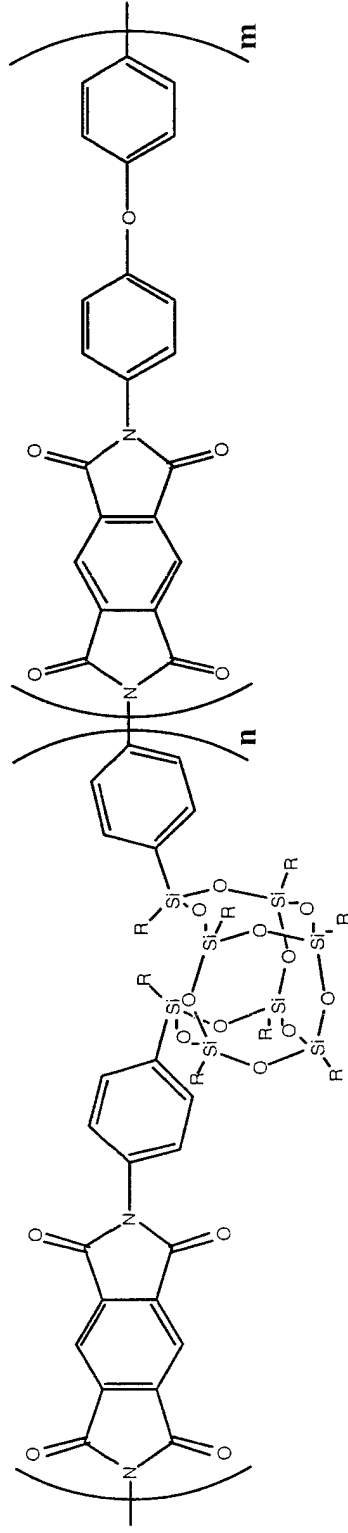
Solid Rocket Motor Insulation



POSS-Insulation Sample



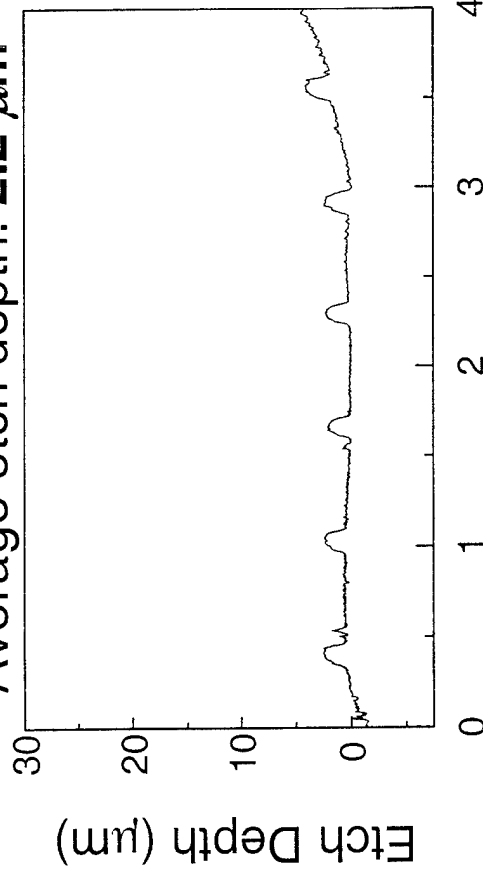
Space Survivable Materials



O atom fluence: 8.47×10^{20} atoms cm^{-2}

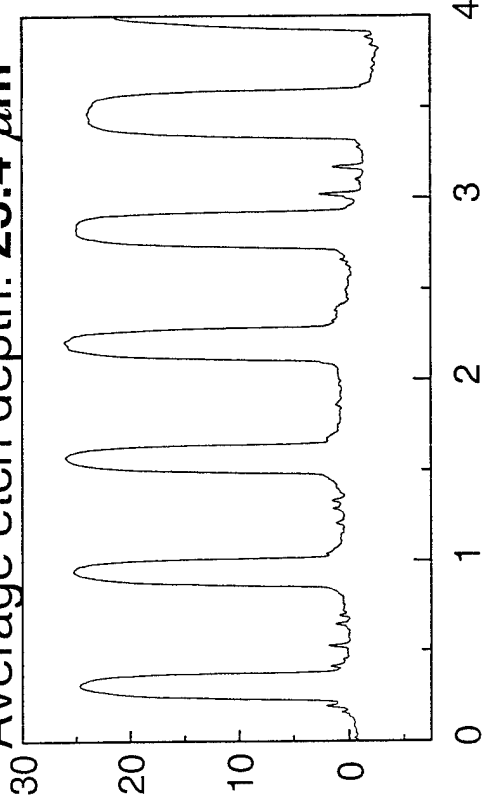
Kapton 10 wt % POSS

Average etch depth: $2.2 \mu\text{m}$



Kapton H Standard

Average etch depth: $25.4 \mu\text{m}$



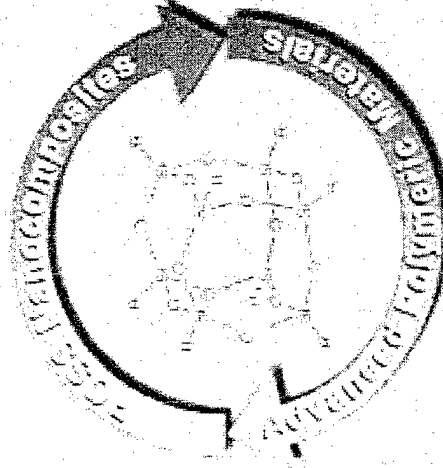
Scanning Length (mm)

Summary

- Nano-sized inorganic clusters (POSS) can be successfully incorporated into a wide variety of different organic polymers.
- These POSS clusters cause increases to the thermal transitions and mechanical properties of the polymers they are copolymerized into.
- Not every POSS is the same and the POSS effect on the properties of analogous polymers shows a dependency on the type of alkyl group on the POSS cluster.
- Rheology of high molecular weight PDMS grafted with small amounts of POSS illustrates a dependence on both the POSS-alkyl-group and POSS shape.
- TEM images of randomly copolymerized polymers illustrate this dependency, as the size of the POSS domains are alkyl-group dependent.

Acknowledgement\$

Capt. Rene Gonzalez
Mr. Brian Moore
Mrs. Becky Morello
Mr. Pat Ruth
Mrs. Sherly Largo
Dr. Darrell Marchant
Dr. Brent Viers
Dr. Rusty Blanski
Dr. Sandra Tomczak
Dr. Shawn Phillips



Prof. Pat Mather UCONN
Prof. Andre Lee MSU
Prof. Ben Hsiao SUNY
Prof. Frank Feher UCI
Prof. Gar Hoflund UF
Prof. Tim Minton MSU

Hybrid Plastics Inc.

Financial \$upport:
Air Force Office of Scientific Research
Air Force Research Laboratory, Propulsion Directorate